
Jitendra kumar\(^1\), N.A Ansari\(^2\), Vikas Verma\(^3\), Sanjeev Kumar\(^4\)

\(^1\) M.Tech Scholar, Department of Mechanical Engineering, Delhi Technological University, New Delhi-110042, India,
\(^2\) Asst. Professor, Department of Mechanical Engineering, Delhi Technological University, New Delhi-110042, India,
\(^3\) Ph.D. Scholar, Department of metallurgical and materials engineering, IIT Roorkee, Uttarakhand, -247667, India
\(^4\) Asst. Professor, Department of Mechanical Engineering, KIMT, U.P Moradabad -244001U.P, India

**Abstract:** It is well known that the future availability of energy resources, as well as the need for reducing \(\text{CO}_2\) emissions from the fuels used has increased the need for the utilization of regenerative fuels. This research is done taking commercial gasoline as reference which is originally blended with 5% ethanol. Hence 5%, 10%, 15%, 20% ethanol blended with Gasoline initially was tested in SI engines. Physical properties relevant to the fuel were determined for the four blends of gasoline. A four cylinder, four stroke, varying rpm, Petrol (MPFI) engine was tested on blends containing 5%, 10%, 15%, 20% ethanol and performance characteristics, and exhaust emissions were evaluated. Even though higher blends can replace gasoline in a SI engine, results showed that there is a reduction in exhaust gases, such as \(\text{HC}, \text{O}_2, \text{CO}, \text{CO}_2\) and increase in Brake Thermal Efficiency on blending. Hence we can conclude from the result that using 10% ethanol blend is most effective and we can utilize it for further use in SI engines with little constraint on material used to sustain little increase in pressure.

**Keywords:** Spark Ignition Engine, Ethanol, Brake Thermal Efficiency, Emission, Gasoline etc.

### I. INTRODUCTION

Rising fuel prices and increased oil consumption along with the lack of sustainability of oil-based fuels have generated an interest in alternative, renewable sources of fuel for internal combustion engines, namely alcohol-based fuels. Currently ethanol is the most widely used renewable fuel with up to 10% by volume blended in to gasoline for regular engines or up to 85% for use in Flex-Fuel vehicles designed to run with higher concentrations of ethanol. Ethanol can also be used as a neat fuel in spark-ignition (SI) engines or blended up to 40% with Diesel fuel for use in compression-ignition (CI) engines [1-2]. Ethanol was introduced as a replacement for methyl tertiary butyl ether (MTBE) when it was realized that MTBE leaked onto the ground at filling stations resulting in the contamination of large quantities of groundwater. Ethanol is biodegradable, less detrimental to ground water, and has an octane number much higher than gasoline as well as having a positive effect on vehicle emissions [3]. There are lots of gases in the environment which are causing pollution and greenhouse effect and the major contributor is the transport sector due to the heavy, and increasing, traffic levels. In spite of ongoing activity to promote efficiency, the sector is still generating significant increases in \(\text{CO}_2\) emissions. As transport levels are expected to rise, especially in developing countries, fairly drastic political decisions may have to be taken to eradicate this problem in the future. Furthermore, the dwindling supply of petroleum. Today, the transport sector is a major contributor to net emissions of greenhouse gases, of which carbon dioxide is particularly important. The carbon dioxide emissions originate mainly from the use of fossil fuels; mostly gasoline and diesel oil in road transportation systems, although some originates from other types of fossil fuels such as natural gas and Liquefied Petroleum Gas (LPG). If international and national goals (such as those set out in the Kyoto protocol) for reducing net emissions of carbon dioxide are to be met, the use of fossil fuels in the transport sector has to be substantially reduced. This can be done, to some extent, by
increasing the energy efficiency of engines and vehicles and thus reducing fuel consumption on a volume per unit distance travelled basis. However, since the total transportation work load is steadily increasing such measures will not be sufficient if we really want to reduce the emissions of carbon dioxide.

1.1 Ethanol as a Blend

In the medium term ethanol produced from grain will probably be the most important alternative fuel for replacing gasoline and in the long term ethanol produced from cellulose might take over from grain ethanol. Today, ethanol accounts for a substantial part of the alternative fuel market. From an international perspective, most research up to 1990 was focused on blends of methanol and gasoline, but some studies were carried out on ethanol-gasoline blends. Since these studies were carried out in the USA, it can be assumed that they mainly included vehicles with efficient emission control systems, but at the same time technical features of cars in the USA have historically differed, at least in part, from those in Sweden. It should also be noted that for a longtime 10% ethanol has been added to commercial gasoline in many parts of the world. In the US there is considerable experience of adding higher proportions of ethanol to gasoline than those allowed by gasoline regulations in Sweden (Europe). The primary advantage of adding a bio based alcohol to gasoline is that it reduces net CO₂ emissions but it also has other positive effects, such as increasing the octane value of the fuel and reducing the benzene content of the exhaust gases. The use of alcohol blended gasoline and neat fuel alcohols as substitutes for neat gasoline have become matters of interest in many countries. The International Energy Agency (IEA), established in 1974, follows the development, and data and other experience from various trials have been presented and discussed at symposia organized by the International Symposium on Alcohol Fuels (ISAF).

1.2 Co-products of ethanol

The co-products that results when making ethanol are dependent on the medium used to produce the ethanol. Table 1 shows a summary of the co-products and what they are used for.

<table>
<thead>
<tr>
<th>S.No.</th>
<th>By-Products/Co-Products</th>
<th>Used For</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Flour, Corn Oil, Corn Meal, Corn Grits</td>
<td>Used in producing foods for human consumption</td>
</tr>
<tr>
<td>2</td>
<td>Fermentation Solids</td>
<td>Used as a high fibre and protein food additive</td>
</tr>
<tr>
<td>3</td>
<td>Corn Gluten Meal and Corn Gluten Feed</td>
<td>Used as high protein animal feed additives</td>
</tr>
<tr>
<td>4</td>
<td>Amino Acids</td>
<td>Used as animal feed additives</td>
</tr>
<tr>
<td>5</td>
<td>Dry Distiller's Grains</td>
<td>Used as high protein and energy animal feed</td>
</tr>
<tr>
<td>6</td>
<td>Carbon Dioxide</td>
<td>Used as a refrigerant, in carbonated beverages, to help vegetables grow more rapidly in greenhouses, and to flush oil wells</td>
</tr>
</tbody>
</table>

In practice, about two-thirds of each tone of grain (i.e., the starch) is converted to ethanol. The remaining by-product is a high protein livestock feed which is particularly well suited for ruminant animals such as cattle and sheep. This by product is also known as Distillers' Dried Grains, DDGS. The protein in this material is utilized more efficiently in ruminant nutrition than are other high-protein feed ingredients such as soybean meal. This by-product of ethanol production is particularly good for Canadian dairy, beef and sheep production. It improves the competitive position globally of producers of these farm commodities. The manure from livestock can be used as a major source of fertilizer in grain crop production. Carbon dioxide is another by-product produced when making ethanol. Carbon dioxide, given off in great quantities during fermentation will be collected and cleaned of any residual alcohol, compressed and sold as an industrial commodity.

II. LITERATURE SURVEY

Fikret Yuksel et al. [4] one of the major problems for the successful application of gasoline–alcohol mixtures as a motor fuel is the realization of a stable homogeneous liquid phase. To overcome this problem, a new carburetor was designed. With the use of this new carburetor, not only the phase problem was solved but also the alcohol ratio in the total fuel was increased. By using ethanol–gasoline blend, the availability analysis of a spark-ignition engine was experimentally investigated. Sixty percent ethanol and 40% gasoline blend was exploited to test the performance, the fuel consumption, and the exhaust emissions. As a result of this study, it was seen that a new dual fuel system could be serviceable by making simple modifications on the carburetor and these modifications would not cause complications in the carburetor system. Ceviz M.A et al. [5] investigated
the effects of using ethanol–unleaded gasoline blends on cyclic variability and emissions in a spark-ignited engine. Results of this study showed that using ethanol–unleaded gasoline blends as a fuel decreased the coefficient of variation in indicated mean effective pressure, and CO and HC emission concentrations, while increased CO₂ concentration up to 10vol. % ethanol in fuel blend. On the other hand, after this level of blend a reverse effect was observed on the parameters aforementioned. The 10vol. % ethanol in fuel blend gave the best results.

Altun Sehmus et al. [6] experimentally investigated the effect of unleaded gasoline and unleaded gasoline blended with 5% and 10% of ethanol or methanol on the performance and exhaust emissions of a spark-ignition engine. The engine tests were performed by varying the engine speed between 1000 and 4000 rpm with 500 rpm period at three fourth throttle opening positions. The results showed that brake specific fuel consumption increased while brake thermal efficiency, emissions of carbon monoxide (CO) and hydrocarbon (HCs) decreased with methanol-unleaded gasoline and ethanol-unleaded gasoline blends. It was found that a 10% blend of ethanol or methanol with unleaded gasoline works well in the existing design of engine and parameters at which engines are operating. Amit Pal et al. [7] operated a Kirloskar, four stroke, 7.35kW, twin cylinder, DI diesel engine in dual fuel mode (with substitution of up to 75% diesel with CNG). The results of this experiment of substituting the diesel by CNG at different loads showed significant reduction in smoke, 10 to 15 % increase in power, 10 to 15 %reduction in fuel consumption and 20 to 40 % saving in fuel cost (considering low cost of CNG). The most exciting result was about 33% reduction in engine noise which may prolong the engine life significantly and the consequent sound levels of giant diesel engine reduced to that of a similarly sized gasoline engine. Hubballi P.A et al. [8] investigated experimentally the effect of Denatured spirit (DNS) and DNS-Water blends as fuels in a four cylinder four stroke SI engine. Performance tests were conducted to study Brake Thermal Efficiency (BThE), Brake Power (BP), Engine Torque (T) and Brake Specific Fuel Consumption (BSFC). Exhaust emissions were also investigated for carbon monoxide (CO), hydrocarbons (HC), oxides of nitrogen (NOₓ) and carbon dioxide (CO₂). The results of the experiments revealed that, both DNS and DNS95W5 as fuels increase BThE, BP, engine torque and BSFC. The CO, HC, NOₓ and CO₂ emissions in the exhaust decreased. The DNS and DNS95W5 as fuels produced the encouraging results in engine performance and mitigated engine exhaust emissions. N. Seshiah et al. [9] tested the variable compression ratio spark ignition engine designed to run on gasoline with pure gasoline, LPG (Isobutene), and gasoline blended with ethanol 10%, 15%, 25% and 35% by volume. Also, the gasoline mixed with kerosene at 15%, 25% and 35% by volume without any engine modifications has been tested and presented the result. Brake thermal and volumetric efficiency variation with brake load is compared. CO and CO₂ emissions have been also compared for all tested fuels. It is observed that the LPG is a promising fuel at all loads lesser carbon monoxide emission compared with other fuels tested. Using ethanol as a fuel additive to the mineral gasoline, (up to 30% by volume) without any engine modification and without any losses of efficiency, it has been observed that the petrol mixed with ethanol at 10% by volume is better at all loads and compression ratios.

III. EXPERIMENTAL SETUP

3.1 Descriptions

The setup consists of four cylinder, four stroke, Petrol (MPFI) engine connected to eddy current type dynamometer for loading. It is provided with necessary instruments for combustion pressure and crank-angle measurements. These signals are interfaced to computer through engine indicator for P-V diagrams. Provision is also made for interfacing airflow, fuel flow, temperatures and load measurement. The set up has stand-alone panel box consisting of air box, fuel tank, manometer, fuel measuring unit, transmitters for air and fuel flow measurements, process indicator and engine indicator. Rotameters are provided for cooling water and calorimeter water flow measurement.

The setup enables study of engine performance for brake power, indicated power, frictional power, BMEP, IMEP, brake thermal efficiency, indicated thermal efficiency, Mechanical efficiency, volumetric efficiency, specific fuel consumption, A/F ratio and heat balance. Windows based Engine Performance Analysis software package “Engine soft” is provided for online performance evaluation.
3.2 Specification

<table>
<thead>
<tr>
<th>S.NO</th>
<th>Equipment</th>
<th>Sub-parts</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Product</td>
<td>Engine test setup 4 cylinder, 4 stroke, Petrol(Computerized)</td>
</tr>
<tr>
<td>2</td>
<td>Product code</td>
<td>233</td>
</tr>
<tr>
<td>3</td>
<td>Engine</td>
<td>Make Maruti, Model Wagon-R MPFI, Type 4 Cylinder, 4Stroke, Petrol (MPFI), water cooled, Power 44.5Kw at 6000 rpm, Torque 59 NM at 2500rpm, stroke 61mm, bore 72mm, 1100 cc, CR 9.4:1</td>
</tr>
<tr>
<td>4</td>
<td>Dynamometer</td>
<td>Type eddy current, water cooled, with loading unit</td>
</tr>
<tr>
<td>5</td>
<td>Propeller shaft</td>
<td>With universal joints</td>
</tr>
<tr>
<td>6</td>
<td>Air box</td>
<td>M S fabricated with orifice meter and manometer (Orifice dia 40 mm)</td>
</tr>
<tr>
<td>7</td>
<td>Fuel tank</td>
<td>Capacity 15 lit with glass fuel metering column</td>
</tr>
<tr>
<td>8</td>
<td>Calorimeter</td>
<td>Type Pipe in pipe</td>
</tr>
<tr>
<td>9</td>
<td>Piezo sensor</td>
<td>Resolution 1 Deg, Speed 5500 RPM with TDC pulse</td>
</tr>
<tr>
<td>10</td>
<td>Crank angle sensor</td>
<td>Input Piezo sensor, crank angle sensor, No of channels 2, Communication RS232</td>
</tr>
<tr>
<td>11</td>
<td>Engine indicator</td>
<td>Type RTD, PT100 and Thermocouple, Type K</td>
</tr>
<tr>
<td>12</td>
<td>Digital milivoltmeter</td>
<td>Range 0-200mV, panel mounted</td>
</tr>
<tr>
<td>13</td>
<td>Temperature sensor</td>
<td>Type two wire, Input RTD PT100, Range 0-100 Deg C, Output 4-20 mA and Type two wire, Input Thermocouple, Range 0-1200 Deg C, Output 4-20 mA</td>
</tr>
<tr>
<td>14</td>
<td>Temperature transmitter</td>
<td>Digital, Range 0-50 Kg, Supply 230VAC</td>
</tr>
<tr>
<td>15</td>
<td>Load sensor</td>
<td>Load cell, type strain gauge, range 0-50 Kg</td>
</tr>
<tr>
<td>16</td>
<td>Fuel flow transmitter</td>
<td>DP transmitter, Range 0-500 mm WC</td>
</tr>
<tr>
<td>17</td>
<td>Airflow transmitter</td>
<td>Pressure transmitter, Range (-) 250 mm WC</td>
</tr>
</tbody>
</table>

Figure 1: Diagram of Setup used
19. Rotameter  Engine cooling 100-1000 LPH; Calorimeter 25-250 LPH

20. Pump  Type Monoblock

21. Add on card  Resolution 12 bit, 8/16 input, Mounting PCI slot

22. Software  “Enginesoft” Engine performance analysis software

IV. RESULT AND DISCUSSION

Gasoline Blends having 5%, 10%, 15% and 20% Ethanol is prepared. Brake thermal efficiency, HC exhaust was plotted in parts per million, O₂, CO, CO₂ were plotted on volume percentage basis. These curves are plotted firstly at no load and then at constant rpm of 3000 and 4000. The density and Lower calorific value of blends are first calculated and then fed in the software set up configuration to get the desired results. The results obtained were noted and then curves were plotted as shown below to have a clear understanding of the variations of different parameters by using different blends.

4.1 No Load Test

![Image of HC exhaust variation with blends at different rpm.](image1)

![Image of O₂ exhaust variation with blends at different rpm.](image2)
Figure 5. CO variation with blends at different rpm.

Figure 6. CO$_2$ variation with blends at different rpm.
At no Load conditions some points are clear which are given below.

- HC emission decreases as blending increases up to 4000 rpm with respect to E5 and is lowest at 2500 rpm. For 10% blend HC emission reduces by 23.08% at 2100 rpm in comparison to commercial Gasoline.
- O₂ Percentage increases as blending increases from 5% and is highest between 2500 rpm to 3500 rpm.
- CO₂ increases up to 4000 rpm when blending increased from 5% and is highest at 2500 rpm. For 10% blend it increases by 0.68% at 2500 rpm in comparison to commercial Gasoline.
- CO decreases as blending is increased and is lowest at 2100 rpm. For 10% blend, it reduces by 35% in comparison to commercial Gasoline.
- Brake Thermal Efficiency increases on blending. Brake Thermal Efficiency reaches a maximum at around 4500 rpm and then starts decreasing. In comparison to commercial Gasoline it increases by 11.6% for 10% blend, 8.1% for 15% blend and 23.37% for 20% blend at 4500 rpm.

4.2 Constant rpm Test

Figure 8. Variation of HC emission with load at 3000 rpm

Figure 9. Variation of HC emission with load at 4000 rpm
**Fig 10.** Variation of $O_2$ emission with load at 3000 rpm

**Fig 11.** Variation of $O_2$ emission with load at 4000 rpm

**Fig 12.** Variation of CO emission with load at 3000 rpm
Figure 13. Variation of CO emission with load at 4000 rpm

Figure 14. Variation of CO$_2$ emission with load at 3000 rpm

Figure 15. Variation of CO$_2$ emission with load at 4000 rpm
At Constant RPM some points are clear which are given below.

- HC emission increases with blending and is more at 3000 rpm compared to 4000 rpm for low loads. At 5kg load, it increases by 6.67% for 10% blend at 3000 rpm and increases by 25% for 10% blend at 4000 rpm with respect to commercial Gasoline.
- CO\textsubscript{2} generally decreases with blending and is generally more for 3000 rpm as compared to 4000 rpm. At 5kg load, it decreases by 2.04% for 10% blend at 3000 rpm and decreases by 2.94% for 10% blend at 4000 rpm with respect to commercial Gasoline.
- CO is less for 3000 rpm as compared to 4000 rpm. At 5kg load, it decreases by 33.33% for 10% blend at 3000 rpm and increases by 18.75% for 10% blend at 4000 rpm with respect to commercial Gasoline.
- O\textsubscript{2} Percentage decreases with blending and is less for 3000 rpm.
- Brake Thermal Efficiency increases on blending. It reaches a maximum at 15 kg load and is generally higher for 3000 rpm than 4000 rpm. At 20kg load, it increases by 45% for 10% blend, 32.2% for 15% blend, 7.91% for 20% blend at 3000 rpm and increase by 39.1% for 10% blend, 17% for 15% blend and 2.99% for 20% blend at 4000 rpm with respect to commercial Gasoline.
V. CONCLUSION

From the results, it can be concluded that Ethanol blends are quite successful in replacing pure Gasoline in Spark Ignition Engine. Results clearly show that there is a decrease in exhaust emissions, increase in Brake Thermal Efficiency. So from the curves it is seen that 10% ethanol blended Gasoline is the best choice for use in the existing Spark Ignition Engines without any modification to reduce exhaust and increase Efficiency. A little consideration has to be taken on material used as maximum pressure inside cylinder is increased by blending. A balance has to be made between Specific Fuel Consumption and Efficiency to take care of users using blend as more fuel will be consumed due to blending of Ethanol with gasoline.

REFERENCES