

in which means:1-full solubility of elements in gold; 2-full insolubility of an element in gold; 3-limitedsolubility of gold in elements; 4-limited solubility of elements in gold, and 5-limited mutual solubility between gold and elementsFurther informations from this kind of diagram cold not be taken.

II. KONCEPT OF ELEMENTARY CELL:

Elementary cell is commonly used in crystallography, metallography or science of materials. The concept of elementary cell is strongly defined by axes x,y,z and their intersections a, b, c and angles between them α , β and γ , as shown in Fig. 2b).

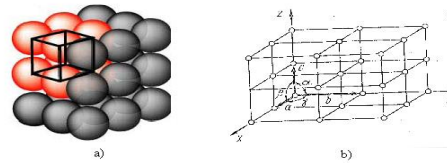


Fig. 2.- Gathering of atoms in crystal a) and elementary cell b)

The crystal classes fall into seven crystal systems, with greater number of unit cells. Every phase has own unit cell.

III. STRUCTURES OF ELEMENTARY CELLS:

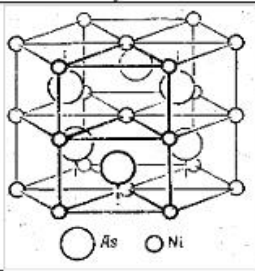
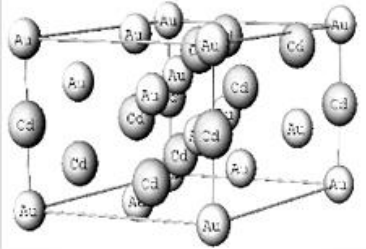
For serios metallurgical analyzing of an alloy, the exact structure of an elementary cell should be known. Those cells for almost gold binary alloys are reviewed in table as follows [4,6].

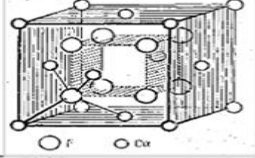
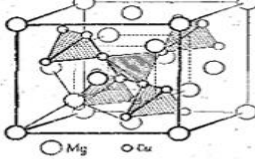
Table 1. Important structures of intermetallic compounds in binary gold alloys

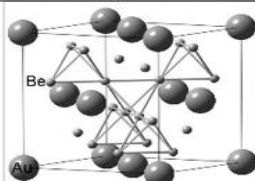
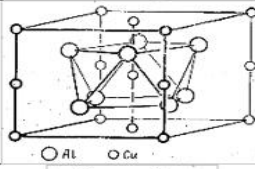
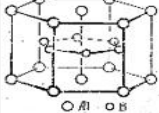
(designation is provided according to International Crystallography Organization)

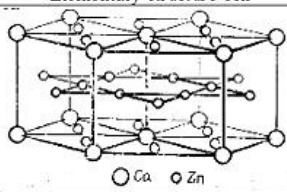

Structure type	Elementary structure cell	Phase
A2 (type W)		β -AlAu ₄
A3 (type W)		ϵ -AuCd α_2 -Au ₃ Cd Au ₃ Ga Au ₃ Hg Au ₄ In ϵ -AuSb β -Au ₄ Sn ϵ -AuZn ₈

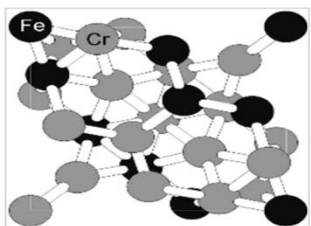
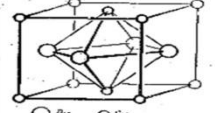
Structure type	Elementary structure cell	Phase
A15 (type Cr ₃ Si)		AuNb ₃ AuTa ₃ AuTi ₃ AuV ₃ AuZr ₃
B2 (type CsCl)		AuCd AuCs AuDy AuEr AuGd AuHo AuLu AuMg AuMn AuNd AuPr AuSc AuSm AuTb AuTm AuY

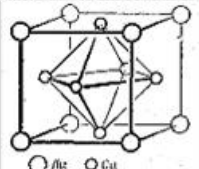
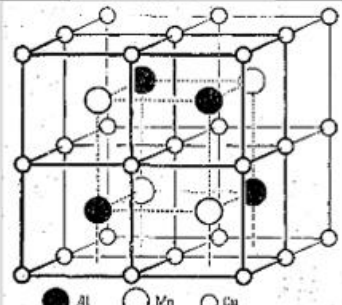
Structure type	Elementary structure cell	Phase
B8 ₁ (type NiAs)		AuSn
B19 (type AuCd)		AuCd AuTi

Structure type	Elementary structure cell	Phase
C1 (type CaF ₂)		Al ₂ Au AuGa ₂ AuIn ₂
C15 (type CsCl)		Au ₂ Bi Au ₂ Na Au ₂ Pb

Structure type	Elementary structure cell	Phase
C15 ₀ (type AuBe ₃)		AuBe ₅ Au ₃ Ca
C16 (type CuAl ₂)		AuNa ₂ AuPb ₂ AuTh ₂
C32 (type AlB ₂)		AuB ₂ Au ₂ Nb Au ₂ Th

Structure type	Elementary structure cell	Phase
D ₂₄ (type CaZn ₅)		Au ₅ Ba Au ₅ K Au ₅ Rb Au ₅ Sr
D ₈₃ (type Cu ₉ Al ₄)		Al ₂ Au ₃ δ-Au ₃ Cd ₈ Au ₃ Ga Au ₃ Hg ₈ Au ₇ In ₃ γ ₁ -AuZn ₃ Au ₃ Zn ₈

Structure type	Elementary structure cell	Phase
D ₈₆ (type CrFe ; σ-phase)		AuTa ₂ AuZr ₃
L ₁₀ (type AuCu)		AuCu

Structure type	Elementary structure cell	Phase
L ₁₂ (type Cu ₃ Au)		Au ₃ Cd AuCu ₃ Au ₃ Cu AuMn ₃ Au ₀ 3300 ₃ Pt Au ₃ Zn
L ₂₁ (type Cu ₂ AlMn)		AgAuCd ₂ AuCuZn ₂ AgAuZn ₂ Au ₅ Sr

NOTE: 1- Structure types are designated according to the most known type, for example AuCu,

CrFe (σ-phase), etc.;

2- Types of phases are listed according to a,b,c ... order

Phases from ternary, quaternary, etc. alloys are pretty complex in their nature, also for investigation and/or structure determination, and obviously they need much more knowledge and/or researching activities [6-12].

IV. DISCUSSION

For most commercial golden alloys with silver and copper the crystal structures were partially investigated, while for many other gold-metal (metalloid) alloys such data still are unknown to producers, even to specialists [1,6,8]. During manufacturing processing, as expected, many different kinds of structures will be appeared, as well in golden products. From the constitutional diagram one can find a valuable data about temperature vs phase composition or solubility of components etc, but not the informations about crystal type structure(s) [4,5]. Only some constitutional diagrams of binary gold alloys are well investigated. The most of phases or intermetallic compounds in gold-metal system also are not fully examined. Such review of crystal structures will be helpful for everyone who is engaged in production or quality assessment of gold alloys [8-12]. The crystal classes fall into seven crystal systems, with greater number of unit cells. A great number of intermetallic compounds which gold makes with other metals here are briefly listed, according to phases between two elements, structure type and type of elementary structure cell. All elements obviously did not make an intermetallic compound, but great number does. Those elementary cells of gold-metal/metalloid here are given for next alloying elements: silver, copper, zinc, nickel, cadmium, tin, bismuth, lead, magnesium, manganese, gallium, sodium, potassium, barium, calcium, mercury, tantalum, vanadium, zirconium, caesium, dysprosium, erbium, gadolinium, holmium, lutetium, neodymium, praseodymium, samarium, rubidium, terbium, thulium, yttrium and strontium. Details about concentration and temperature regions of those intermetallic compounds could be found at a proper constitutional diagram. For detailed structure determinations of phases from gold-metal(s) systems different methods of investigations should be applied [2,6,12].

V. CONCLUSION

The elementary cells for almost gold alloys are reviewed, where the crystal types are shown for every group of cell, mainly for binary systems, for main elements which are able to make an intermetallic compound with gold. The designation of important structures, i.e. crystallographic cells of intermetallic compounds in binary gold alloys, is provided according to International Crystallography Organization. Here collected and shown data will be helpful for everyone who is engaged in structure determination or sometimes in production of such golden alloy. However, the obtained results should be used for further explanations of physical, chemical or metallurgical properties of gold alloys, neither only for jewelry purposes.

REFERENCES

- [1]. В.М. Малишев, Д.В. Румјанцев: Золото, Москва 1979, Металлургија, стр. 212- 221.
- [2]. [C.S. Barret, T.B. Massalski: Structure of Metals, перевод с англискогo, Москва 1984, Металлургија, Част I, стр. 15-45.
- [3]. R.E. Smallman: Modern Physical Metallurgy, London 1970, Butterworth, pp. 140-165.
- [4]. G. Schulze: Metallphysik, перевод с немецкогo, Москва 1971, МИР, стр. 37-93; 430-459.
- [5]. L.S. Dent Glasser: Crystallography and its Applications, New York 1977, Van Nostrand Reinhold Co. Ltd, pp. 3-59.
- [6]. З. Карастојковић: Докторска теза, Центар за мултидисциплинарне студије, Универзитет у Београду, Београде 2013.
- [7]. М.И. Голјдштејн, В.С. Литвинов, В.М. Бронфин: Металлофизика високопрочних сплавов, Москва 1986, Металлургија, стр. 12-76. R. Süss, E.v.d. Lingen and L. Glaner, 18 CARAT YELLOW GOLD ALLOY WITH INCREASED HARDNESS, PHYSICAL METALLURGY DIVISION, Mintek, Priv.Bag x 3015, Randburg 2125, South Africa, Gold Bulletin, (2004), 3713-4, p.p. 196-207;
- [8]. H. Okamoto, D.J. Chacrabarti, D.E. Janghlin & Massalski, THE Au-Cu (GOLD – COOPER) SYSTEM, Bulletin of Alloy Phase Diagrams, Vol.8, No. 5, (1987), p.p.454-473;
- [9]. R. Perić, Z. Karastojković, Z. Kovačević, D. Gusković, THE AGE-HARDENING MECHANISM OF Au585Ag90Cu264Zn61 GOLD ALLOY FOR JEWELRY PRODUCTION, III Int.Congres, Jahorina, R.Srpska, Proceedings, CD issue, paper M-33, pp. 951-960.
- [10]. R. Perić, Z. Karastojković, Z. Kovačević, D. Gusković, HARDNESS CHANGES IN Ag925Cu75 JEWEL ALLOY AFTER TEMPERING OF AS-CASTED, ROLLED AND QUECHED STATES, 43rd IOC on Mining and Metallurgy, Proceedings, Kladovo (2011), p.p. 171-174.
- [11]. Z. Karastojković, R. Perić, Z. Janjušević, N. Bajić: SPECIFIC ELECTRICAL CONDUCTIVITY MEASUREMENTS FOR QUALITY ASSESSMENT OF GOLD ALLOY 585 WITH PALLADIUM, 8th Int. Quality Conf, May 2014, Kragujevac, pp. 477-480.

Z. Karastojković "Review of Elementary Crystallographic Cells of Important Intermediate Phases in Golden Alloys" American Journal of Engineering Research (AJER), vol. 7, no. 2, 2018, pp. 126-130.